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Measure Amounts Of A Substance Because Chemical Reactions Often Involve Large Numbers Of Small Particles. 10. Circle The Letter Of The Correct Answer. If One Carbon Atom Has An Atomic Mass Of 12.0 Amu And One Oxygen Atom Has An Atomic Mass Of 16.0 Amu, What Is The Molar Mass Of Carbon Dioxide? A.28.0 Amu B.44.0 Amu C.28.0 G D.44.0 G 11. May 3th, 2024

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3) $2(aq) + Y$ C. $K(s) + H_2SO_4(aq) + Y$ D. $Zn(s) + NaOH(aq) + Y$ SECTION 11.3 REACTIONS IN AQUEOUS SOLUTION 1. Write The Net Ionic Equation For The Reaction Between Aqueous Barium Nitrate, $Ba(NO_3)_2$, And Sodium Sulfate, Na_2SO_4 . 2. Magnesium Reacts With HCl To Form Hydrogen And Magnesium Chloride. Write The Balanced Net Ionic Equation For This Reaction. 3. May 2th, 2024

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(b) This Is A Chemical Reaction As A Gas Is Created (carbon Dioxide) During The Reaction. (a) + Hacog (b) The Bottle Could Be Heated Up Or Opened To Reverse

The Reaction. CHECK YOUR LEARNING Suggested Answers 1. The Arrow Shows Which Way The Chemical Reaction Is Going. 2. May 3th, 2024

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POGIL 03 - Chemical Reactions 8 - Combustion Reactions

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