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Involved In A Chemical Reaction Are Called Reactants,

But The Newly Formed Substances Are Called The

Products. The Products Are New Substances With

Properties That are Different From Those Of Reactants

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Chapter 10 (Remember The Following Are Diatomic: H
2, N 2, O 2, F 2, Cl 2, Br 2, I 2) The Coefficients Should
Add Up To The Number At The End That Is In
Parenthesis. 1. Carbon + Oxygen Carbon Dioxide (0) 2.
Copper + Silver Nitrate Copper(II) Nitrate + Silver (4)
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Balancing Equations. 1. Word Equation: Nitrogen Gas
Reacts With Hydrogen Gas To Form Ammonia Gas.
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Chemistry 11 - Balancing Chemical Equations
1. Balance The Following Equations
A. $\text{NH}_3 + \text{O}_2 \rightarrow \text{NO} + \text{H}_2\text{O}$
B. $(\text{NH}_4)_2\text{C}_2\text{O}_4 + \text{AlCl}_3 \rightarrow \text{Al}_2(\text{C}_2\text{O}_4)_3 + \text{NH}_4\text{Cl}$
C. $\text{C}_{13}\text{H}_{28} + \text{O}_2 \rightarrow \text{CO}_2 + \text{H}_2\text{O}$
D. $\text{Fe} + \text{H}_2\text{SO}_4 \rightarrow \text{F}$
Apr 2th, 2024.

Writing And Balancing Chemical Equations Chemistry 110
Writing And Balancing Chemical Equations Chemistry 110
I] On The Line At The Left, Write The Letter Corresponding To The Reaction Type: (A) Combination/Synthesis (B) Decomposition (C) Replacement/single Replacement Jun 1th, 2024

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