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** Acids: Start With "H" Or End In $-CO_2H$ ($-COOH$) And Have A Sour Taste. ** Bases:

End In -OH (positive Ion In Front) Or End In -NH, -NH₂, NH₃. They Have A Bitter Taste And Feel Slippery. * 3 Acid/Base Theories . I. Arrhenius Theory A) Acids Produce H⁺, 2024

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Chapter 15: Acids And Bases Arrhenius Definitions: ... In Chapter 4 We Defined Weak Acids And Weak Bases As Weak Electrolytes (only Partially Ionized In Aqueous Solution). ... Calc 1th, 2024

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