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Inorganic Chemistry 1. 1.00g Of Carbon Is Combusted

In A Limited Supply Of Pure Oxygen. 0.50g Of The

Carbon Combusts To Form CO₂ And 0.50 G Of The

Carbon Combusts To Form CO. The Resultant Mixture

Of CO₂ And CO Is Passed Through Excess Na Feb 1th, 2024
Answer On Question #58182 Chemistry Organic Chemistry ...
2) Aniline And Acetanilide Stand In A One-to-one Molar Relationship, So 0.0322 Mol Of Aniline Produces 0.0322 Mol Of Acetanilide. The Amount Of Acetic Anhydride Does Not Play A Role, Since It Is In Excess. Then, $n(\text{N}) = 0.0322$.aniline N Aceta II. Ide Mol We Find The Molar Mass Of Acetanilide: $(\text{N}) (\text{O}) 12 \ 8 \ 1 \ 9 \ 14 \ 1 \ 16 \ 1 \ 135(\text{O}) \cdot 89 \ \text{M Aceta} \dots$ Mar 3th, 2024
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