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[Type Text] [Type Text] [Type Text] LESSON 1 SOME BASIC CONCEPT OF CHEMISTRY Chemistry 1) 40 ml of Methane is completely burnt using 80 ml of Oxygen at room temperature. The volume of gas left after cooling to room temperature is (a) 40 ml CO₂ gas (b) 40 ml CO₂ gas and 80 ml H₂O gas (c) 60 ml CO₂ gas and 60 ml H₂O gas (d) 120 ml CO₂ gas Jul 3th, 2024

UNIT 1 SOME BASIC CONCEPTS OF CHEMISTRY Chemistry

2 Prefixes in SI System Multiple Prefix Symbol 10⁻¹² Pico P 10⁻⁹ Nano N 10⁻⁶ Micro μ 10⁻³ Milli M 10⁻² Centi C 10⁻¹ Deci D 10 Deca Da 10² Hecto H 10³ Kilo K 10⁶ Mega M 10⁹ Giga G 10¹² Tera T Mass and Weight-- Mass of a substance is the amount of matter present in it while weight is the force exerted by gravity on an object. The mass of a substance is File Size: 430KB Jun 3th, 2024

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Of Organic Chemistry, with an emphasis of topics of relevance to Biochemistry. It should be noted that this course does NOT satisfy the requirements for B.S. Chemistry majors (who are required to take CHEM 30481), but is intended for students majoring in Biology, Nutrition, B.A. Chemistry Program, and other related disciplines. Mar 2th, 2024

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In which any number can be represented in the form $N \times 10^x$. N (where N is an exponent having positive or negative values and N can vary between 1 to 10). E.g. We can write 232.508 as 2.32508×10^2 . In scientific notation. Similarly, 0.00016 can be written as 1.6×10^{-4} . Precision refers Apr 3th, 2024

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Balancing Chemical Equations. The following four steps can be used as a guide to balance chemical equations. Example: Sulfuric acid reacts with sodium hydroxide to create sodium sulfate and water. Balance this chemical reaction. Step 1: Write an unbalanced equation, U May 1th, 2024

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