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## **LESSON 1 SOME BASIC CONCEPT OF CHEMISTRY Chemistry**

[Type Text] [Type Text] [Type Text] LESSON 1 SOME BASIC CONCEPT OF CHEMISTRY Chemistry 1) 40 ml of Methane is completely burnt using 80 ml of Oxygen at room temperature. The volume of gas left after cooling to room temperature is (a) 40 ml CO<sub>2</sub> gas (b) 40 ml CO<sub>2</sub> gas and 80 ml H<sub>2</sub>O gas (c) 60 ml CO<sub>2</sub> gas and 60 ml H<sub>2</sub>O gas (d) 120 ml CO<sub>2</sub> gas 3th, 2024

## **UNIT 1 SOME BASIC CONCEPTS OF CHEMISTRY Chemistry**

2 Prefixes in SI System Multiple Prefix Symbol 10<sup>-12</sup> Pico P 10<sup>-9</sup> Nano N 10<sup>-6</sup> Micro  $\mu$  10<sup>-3</sup> Milli M 10<sup>-2</sup> Centi C 10<sup>-1</sup> Deci D 10 Deca Da 10<sup>2</sup> Hecto H 10<sup>3</sup> Kilo K 10<sup>6</sup> Mega M 10<sup>9</sup> Giga G 10<sup>12</sup> Tera T Mass and Weight-- Mass of a substance is the amount of matter present in it while weight is the force exerted by gravity on an object. The mass of a substance is File Size: 430KB 2th, 2024

## **Chemistry 20481 - Basic Organic Chemistry I - Fall 2019**

Of Organic Chemistry, with an emphasis of topics of relevance to Biochemistry. It should be noted that this course does NOT satisfy the requirements for B.S. Chemistry Majors (who are required to take CHEM 30481), but is intended for students majoring in Biology, Nutrition, B.A. Chemistry Program, and other related disciplines. 1th, 2024

## **CBSE Class 11 Chemistry Some Basic Concepts Of Chemistry**

In which any number can be represented in the form  $N \times 10^x$ . N (where N is an exponent having positive or negative values and N can vary between 1 to 10). E.g. We can write 232.508 as  $2.32508 \times 10^2$ . In scientific notation. Similarly, 0.00016 can be written as  $1.6 \times 10^{-4}$ . Precision refers 3th, 2024

## **Chemistry 151: Basic Chemistry Chemical Equations**

Balancing Chemical Equations. The following four steps can be used as a guide to balance chemical equations. Example: Sulfuric acid reacts with sodium hydroxide to create sodium sulfate and water. Balance this chemical reaction. Step 1: Write an unbalanced equation, U 3th, 2024

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