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TABLE OF CONJUGATE ACID-BASE PAIRS Acid Base  $K_a$  (25 OC)  $HClO_4$   $ClO_4^-$  -  $H_2SO_4$   $HSO_4^-$  -  $HCl$   $Cl^-$  -  $HNO_3$   $NO_3^-$  -  $H_3O^+$   $H_2O$   $H_2CrO_4$   $HCrO_4^-$  -  $1.8 \times 10^{-1}$   $H_2C_2O_4$  (oxalic Acid) Jul 2th, 2024

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Acid Base Conjugate Conjugate . Acid Base . 2) What Is The Strongest Base In The Following Reaction?  $HNO_3(aq) + H_2O(l) \rightleftharpoons NO_3^-(aq) + H_3O^+(aq)$   $H_2O$  Is The Strongest Base. Strong Acids, Such As  $HNO_3$  Have Weak Conjugate Bases, So  $NO_3^-$  is A Weak Base. H 2O And Mar 2th, 2024

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Carboxylic Acids Protonated Ketone-7.3 6.37 7 Carbonic Acid Tonic Acid -0.6 Protonated Pyridine 5.2 PKa Chart Conjugate Acid Conjugate Base Conjugate Acid Conjugate Base S T R O N G E S T A C I D S W E A K E S T B A S E S Hydrogen Sulfi May 2th, 2024

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In Fact, When A Wine Has "gone Off" And Has Acquired A Sour Taste, This Is Due To The Oxidation Of The Ethanol In The Wine To Acetic Acid. (The "corking" Of Wine, I.e. Tainting Of The Wine By Compounds Transferred From Or Through The

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