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Sodium Acetate Is Also Used In Heating Pads, Hand Warmers, And Hot Ice. Sodium Acetate Trihydrate Crystals Melt At  $136.4\text{ }^{\circ}\text{F}/58\text{ }^{\circ}\text{C}$  (to  $137.12\text{ }^{\circ}\text{F}/58.4\text{ }^{\circ}\text{C}$ ), Dissolving In Their Water Of Crystallization. When They Are Heated Past The Melting Point And Subsequently Allowed To Cool, The Aqueous Solution Becomes Supersaturated. This Solution 2th, 2024

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67 PREPARATION AND TESTING OF BUFFER SOLUTIONS. P. URPOSE. The Purpose Of The Laboratory Investigation Is To Experimentally Determine (1) PKa (and Thus Ka) Of The Acid In A Buffer And Thus The Buffer Range, (2) Investigate The Buffer

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### **Section 19.1. Acid-Base Buffer Solutions**

If We Add  $\text{CH}_3\text{COO}^-$ -SHIFT Net Effect:  $[\text{H}^+]$  Decreases PH Increases Also,  $[\text{OH}^-]$  Increases POH Decreases I.e., a Weak Acid And Its Conjugate Base Or A Weak Base And Its Conjugate Acid. 19-2 Example: Addition Of  $\text{NH}_4\text{Cl}$  To  $\text{NH}_3$  3th, 2024

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