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Page III-13-1 / Chapter Thirteen Lecture Notes Page III-13-1 / Chapter Thirteen Lecture Notes. The Equilibrium Constant. For Any Type Of Chemical Equilibrium Of The Type.  $A + B \rightleftharpoons C + D$ .  $K_c = \frac{[C][D]}{[A][B]}$ . ... Le Chatelier's Principle, Catalysts Changes In Equilibria.  $K_c = \frac{[C][D]}{[A][B]}$ . Equ Apr 1th, 2024

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