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(c) (i) Write A Chemical Equation Representing The Lattice Energy Of AgBr. (ii) Use
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Ionisation Energy Of Silver = $+731\text{kJmol}^{-1}$ Electron Affinity Of Bromine = 325kJmol^{-1}
Enthalpy Change Of Atomisation Of Silver = $+285\text{kJmol}^{-1}$ Enthalpy Change Of

Atomisation Of Bromine = +112kJmol⁻¹ ... May 2th, 2024

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