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Answer On Question #60099, Chemistry / Inorganic Chemistry 1. 1.00g Of Carbon Is Combusted In A Limited Supply Of Pure Oxygen. 0.50g Of The Carbon Combusts To Form  $\text{CO}_2$  And 0.50 G Of The Carbon Combusts To Form CO. The Resultant Mixture Of  $\text{CO}_2$  And CO Is Passed Through Excess Na 4th, 2024

### **Answer On Question #58182 Chemistry Organic Chemistry ...**

2) Aniline And Acetanilide Stand In A One-to-one Molar Relationship, So 0.0322 Mol Of Aniline Produces 0.0322 Mol Of Acetanilide. The Amount Of Acetic Anhydride Does Not Play A Role, Since It Is In Excess. Then, N ( N ) 0.0322 .aniline N Aceta II. Ide Mol We Find The Molar Mass Of Acetanilide: ( N ) ( ) 12 8 1 9 14 1 16 1 135( ). 89

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