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5. List The Seven Diatomic Molecules. Chapter 9: Chemical Names & Formulas 1. Determine Whether The Following Compounds Are Ionic Or Covalent, And

Then Name Them. CaSO_4 , N_2O_4 , NH_4NO_3 , CaS , Li_3PO_4 , NH_3 , $\text{Cu}(\text{OH})_2$. 2. Complete The Following Table And Name Each Ionic Compound Formed. + Mar 2th, 2024

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CHAPTER 7 REVIEW: REFER TO THE STUDY GUIDE ON PAGE 207 OF TEXT Complete/review The Practice Problems (throughout And At The End Of The Chapter) In The Book And Check Your Answers In The Back (Appendix E, Page R87 - R88) O #s 1 & 2 On P. 193 O #s 12 & 13 On P. 196 O P. 207, #s 30- 83 (answers To Odd Questions Are Given In Appendix) May 3th, 2024

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C. Ag-107, Au-197 And Ar-40 11. List The Order That You Fill The Energy Levels And Orbitals For Electron Configuration. 12. Fill In The Orbital Diagram Below For Tungsten And Then Write The Full Electron Configuration Below. W: 13. Which Of The Following Is An Incorrect Orbital Diagram? Why? Apr 2th, 2024

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18. Write The Abbreviated (noble Gas) Electron Configuration For The Following Atoms: A. Silver B. Copper C. Tungsten D. Titanium 19. Define Ground State. 20. Define Excited State. 21. How Do Atoms Jump From Ground State To Excited State And Back? (Remember Our Flame Test L Jul 2th, 2024

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2. How Does Calcium Obey The Octent Rule When Reacting To Form Compounds> A. It Gains Electrons B. It Gives Up Electrons C. It Does Not Change Its Number Of Electrons D. Calcium Does Not Obey The Octet Rule 3. What Is The Formula When Potassium And Sulfur Combine? K. 2S 4. What Is The F Jun 2th, 2024

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