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Chemistry 11 Stoichiometry Review Worksheet

Name: \_\_\_KEY\_\_\_ 33.  $\text{KClO}_4 \rightarrow 2\text{O}_2 + \text{KCl}$

Actual yield = 1.45g Theoretical yield = 5.95g  $\text{KClO}_4 \times 1\text{mol}$

$122.55\text{g} \times \frac{20}{2} \frac{1}{32.00\text{g}} = 3.11\text{g O}_2$

% Yield =  $\frac{1.45\text{g}}{3.11\text{g}} \times 100\% = 46.6\%$  34.  $4\text{FeCO}_3 + \text{O}_2$

$\rightarrow 2\text{Fe}_2\text{O}_3 + 4\text{CO}_2$

Percent purity = 62.8%? pure  $\text{FeCO}_3$  Jun 2th, 2024

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## **R. Janssen ...**

In A Camping Stove? 2. How Many Moles Of O<sub>2</sub> Molecules Should Be Supplied To Burn 1 Mol Of CH<sub>4</sub> ... Was At One Time Considered For Use As A Rocket Fuel. ... How Many Formula Units Of Calcium Oxide (CaO) May 2th, 2024

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11.4.1 Is Shown In Figure 12.2.1. We Can Use The



Balanced Chemical Equation For The Reaction And  
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44. What Number Of Moles Of  $O_2$  Is Needed To  
Produce 14.2 Grams Of  $PO_4^{3-}$  From  $P_4O_{10}$ ? (Molar Mass  $P_4O_{10} = 284$ ) (A) 0.0500 Mole (B) 0.0625 Mole (C)  
0.125 Mole Feb 3th, 2024

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