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Chemistry 11 Stoichiometry Review Worksheet Name: ___KEY___ 33. $KClO_4 \rightarrow 2O_2 + KCl$ Actual yield = 1.45g Theoretical yield = 5.95g $KClO_4 \times 1 \text{ mol } 122.55 \text{ g} \times 2 O_2 \times 1 KClO_4 \times 32.00 \text{ g } 1 \text{ mol} = 3.11 \text{ g } O_2$ %Yield = $\frac{1.45 \text{ g } 3.11 \text{ g}}{3.11 \text{ g}} \times 100\% = 46.6\%$ 34. $4FeCO_3 + O_2 \rightarrow 2Fe_2O_3 + 4CO_2$ Percent purity = 62.8%? g pure $FeCO_3$ Jun 1th, 2024

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