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A) [Co(H₂O)₆]²⁺ Co +2 E) [FeO₄]²⁻ Fe +6 I) K₂[CoCl₄] Co +2 B) [CrCl₆]³⁻ Cr +3 F) [Mn(CN)₆]⁴⁻ Mn +2 J) K₃[AuF₆] Au +3 C) [Co(NH₃)₆]Cl₂ Co +2 G) [Ni(CO)₄] Ni 0 K) (NH₄)₂[IrCl₆] Ir +4 D) [Co(NH₃)₅]Cl₂ Co +3 H) [Ni(EDTA)]²⁻ Ni +2 L) Na[Mn(CO)₅] Mn -1 TASK 2 - Writing Half Equations A) 2I⁻ → I₂ + 2e⁻ - B ... Mar 3th, 2024

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Metal Carbonate → Metal Oxide + Carbon Dioxide (on Heating) CaCO₃ → CaO + CO₂ TASK 4 - WRITING BALANCED EQUATIONS 1) Balance The Following Equations. Apr 3th, 2024

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