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## 2 Classifying 2 Classifying Organisms Organisms

Why Do Scientists Classify? Just As Shopping Can Be A Problem In A Disorganized Store, fin D-ing Information About A Specific Organism Can Also Be A Prob-lem. So Far, Scientists Have Identified More Than One Million Kinds Of Organisms On Earth. That's A Large Number, And It Is Continually Growing As Scientists Discover New Organisms. Jan 3th, 2024

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### Station #1: Classifying Reactions

Station #5: Classifying & Predicting Products Classify The Following As SD Or DD. Predict The Products. If No Reaction Will

Occur, Write "NR" DD PbCl 2 + 2 AgNO 3 Pb(NO 3) 2 + 2AgCl SD H 2 SO 4 + Fe H 2 + FeSO 4 SD Mg 3 (PO 4) 2 + H 2 NR SD L 2 + KBr NR DD Li 2 SO 4 + Pb(NO 3) 2 2LiNO 3 + PbSO 4 Station #6: Factors That Affect Reaction Jun 2th, 2024

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Concept Review Classifying Chemical Reactions Answer Key Title: Concept Review Classifying Chemical Reactions Answers Author: Wiki.ctsnet.org-Tanja Hueber-2020-09-03-06-08-42 Subject: Concept Review Classifying Chemical Reactions Answers Concept Review Classifying Chemical Reactions Answers The Concept Review. Terms In This Set (66) Chemical ... Jul 4th, 2024

## **Section 3 Classifying Chemical Reactions**

Decomposition Decomposition Reaction- One Substance Breaks Down, Or Decomposes, Into Two Or More Substances. AB  $\rightarrow$  A + B Most Decomposition Reactions Require The Use Of Heat, Light, Or Electricity. Jul 1th, 2024

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Classifying Chemical Reactions Types Of Chemical Reactions: 1. Synthesis 2. Decomposition 3. Combustion 4. Single-Replacement 5. Double-replacement \*Classifying How Atoms Rearrange During A Chemical Reaction. December 12, 2014 1. Synthesis Reaction: 2 Or More Substances React To Produce A Single Product A + B AB Mar 3th, 2024

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#### Section 8-3: Classifying Chemical Reactions

Section 8-3: Classifying Chemical Reactions Reaction Type General Format Characteristics Of Reactants Characteristics Of Products Example (s) Decomposition AB A + B Only One Reactant Two Or More Simpler Products CuCO 3 (s) CuO + CO 2 (g) 2 H 2 O (l) 2 H 2 (g) + O 2 (g) Synthesis AKA: Combination O (l) A + B AB 2 Two Elements Or Two Simple Apr 4th, 2024

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