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$H_2O + CO \rightleftharpoons H_2 + CO_2$ Suppose You Were To Start The Reaction With Some Amount Of Each Reactant (and No H

Jan 4th, 2024 Static Equilibrium For Forces Static Equilibrium And G GGG ... $F_{pivot} = (m_B + m_1 + m_2)g$

$F_{pivot} - m_B g - N_{B,1} - N_{B,2} = 0$ Worked Example: Solution Pivot Force: Lever Law: Pivot $F = (m_B + m_1 + m_2)g$

$= (2.0 \text{ Kg} + 0.3 \text{ kg} + 0.6 \text{ Kg})(9.8 \text{ M} \cdot \text{s}^{-2}) = 28.4 \text{ N}$

$d_1 m_1 = d_2 m_2$ $D_2 = d_1 m_1 / m_2 = (0.4 \text{ M})(0.3 \text{ Kg} / 0.6 \text{ Kg}) = 0.2 \text{ M}$ Generalized Lever Law , , 1 11 22, 2, $\perp \perp = + = +$ FF F FF F & & GG G GGG Feb 2th, 2024.

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