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CHEMICAL SUBSTANCES
AlF₄Na Sodium
Tetrafluoroaluminate -1869.0 -1827.5 345.7 105.9 AlH
Aluminum Hydride 259.2 231.2 187.9 29.4 AlH₃
Aluminum Hydride -46.0 30.0 40.2 AlH₄K Potassium
Aluminum Hydride -183.7 AlH₄Li Lithium Aluminum
Hydride -116.3 -44.7 78.7 83.2 AlH₄Na Sodium
Aluminum Hydride -15.5 Al Aluminum Monoiodide 65.5
36.0 Feb 1th, 2024 Problem Solutions Of Chemical
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Chemistry Webbook, Molecular Formula Crystal Liquid

Gas Name F H ° KJ/mol F G ° KJ/mol S ° J/mol K C P
 J/mol K F H ° KJ/mol F G ° KJ/mol S ° J/mol K C P J/mol K
 F H ° KJ/mol F G ° KJ/mol S ° J/mol K C P J/mol K
 Substances Not Containing Carbon: Ac Actinium 0.0
 56.5 27.2 406.0 366.0 188.1 20.8 Ag Silver 0.0 42.6 ...
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 Equilib- ...1. Examples of Free Energy Calculations Free
 Energy Calculations Are Carried Out In Much The Same
 As Enthalpy calculations. There Are Tables Of Standard
 Free Energies Of Formation Of Compounds. Elements In
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 Free Energy Of Formation. Consider The Following
 Chemical Reaction: $H_2(g) + Cl_2(g) \rightarrow 2HCl(g)$. Jan 3th,
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 Descriptors Of ... Standard Enthalpy Of Formation And
 Energy Of HOMO. The Second PC Is Strongly Influenced
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 Parameters, Number Of Chlorine Atoms At α -positions
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 Properties. Part ... Other than The standard state,
 Value of ΔH_f° represents the apparent Enthalpy Of the
 Reaction Of Formation Of the Solution From The
 Elements Comprising The solute, ... Jan 2th, 2024.
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