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A) [Co(H 2O) 6] 2+ Co +2 E) [FeO 4] 2- Fe +6 I) K 2[CoCl 4] Co +2 B) [CrCl 6] 3- Cr +3 F) [Mn(CN) 6] 4-Mn +2 J) K 3[AuF 6] Au +3 C) [Co(NH 3) 6]Cl 2 Co +2 G) [Ni(CO) 4] Ni 0 K) (NH 4) 2[IrCl 6] Ir +4 D) [Co(NH 3) 5Cl]Cl 2 Co +3 H) [Ni(EDTA)] 2- Ni +2 L) Na[Mn(CO) 5] Mn -1 TASK 2 - Writing Half Equations A) $2l- \rightarrow l 2 + 2e - B \dots 2th, 2024$

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Metal Carbonate \rightarrow Metal Oxide + Carbon Dioxide (on Heating) CaCO 3 \rightarrow CaO + CO 2 TASK 4 - WRITING BALANCED EQUATIONS 1) Balance The Following Equations. 3th, 2024

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3. Using ACD Labs 12.02, Calculate And Report The 13C NMR Chemical Shifts (, Ppm) For Each Of The Following Compounds Below. 4. Using ACD Labs 12.02, For Each Calculated 13C NMR Spectrum In Part 3), Using The "Tools" Menu, Select "Off-Resonance", And Report The Splitting Of Ea 2th, 2024

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Used In Nuclear Magnetic Resonance Spectroscopy. 2.
NMR Theory (13.3-13.5) A. All Nuclei With Unpaired
Protons Or Neutrons Are Magnetically Active- They
Have A Magnetic Field Arising From The Unpaired
Nuclear Particle. Of Greatest Interest To An Organic
Chemist Is Hydrogen (including Deuterium) And
Carbon (The 13C Isotope Not The 12C Isotope ... 4th,

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Chapter 13: Nuclear Magnetic Resonance (NMR)
Spectroscopy Direct Observation Of The H's And C's Of
A Molecules Nuclei Are Positively Charged And Spin On
An Axis; They Create A Tiny Magnetic Field + + Not All
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Important NMR Active Nuclei In Organic Chemistry
Natural Abundance 1H 99.9% 13C 1.1% 1th, 2024

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-When All The Magnetic Spins Are Added Together, There Is A Net Excess Of Spins Aligned With The Field (mm = -1/2) Compared To Against The Field (mm = +1/2) 10 •Due To Thermal Motions, The Difference In Populations Is Quite Small •For Our Magnet N -1/2 N +1/2 =1.000064 For 1H A •The 3th, 2024

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