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13.1 Kinetic Theory And A Model For Gases The Word Kinetic

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1. Describe How Gases, Liquids, And Solids Compare Using The Following Table. Solids Liquids Gases
Volume (definite Or Indefinite) Molecular Motion (high, Med, Low) Distance Between Molecules (g 2. Read Page 420: What Is Kinetic Energy? Describe Kinetic

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CHAPTER 12 GASES AND KINETIC-MOLECULAR THEORY

3 Boyle's Law: The Volume-Pressure Relationship $\{V \propto 1/P$ Or $\{V = K (1/P)$ Or $PV = K\{P_1V_1 = K_1$ For One Sample Of A Gas. $\{P_2V_2 = K_2$ For A Second Sample Of A Gas. $\{k_1 = K_2$ For The Same Sample Of A Gas At The Same T. $\{$ Thus We Can Write Boyle's Law Mathematically As $P_1V_1 = P_2V_2$ 2th, 2024

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Kelvin Is An Absolute Scale. D) There Is No Difference Whatsoever Between Them. 20. What Is The Difference Between The 1th, 2024

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Converting Kelvin To Celsius - 273 Convert 393K To $^{\circ}\text{C}$ - $273 = 20^{\circ}\text{C}$ Converting Fahrenheit To Celsius (not Used Much) Use $C = 5/9(F - 32)$ Example: Convert 80°F To $^{\circ}\text{C}$ $C = 5/9(F - 32) = 26.7^{\circ}\text{C}$. Physics 20: Heat Teacher Notes Page 8 Of 24 Practice Problems $K = C + 273$ $C = K - 273$ 1. Convert Th 1th, 2024

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