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1. NAMING BINARY ACIDS: The Name Of The Binary Acid Consists Of Two Words. The First Word Has Three Parts: The "hydro" Prefix The Root Of The Nonmetal Element The "ic" Ending The Second Word Is Always "acid" Examples: HCl = Hydro Chlor Ic Acid = Hydrochloric Acid HBr = Hydro Brom Ic Acid = Hydrobromic Acid 1th, 2024

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Worksheet 5 - Naming Ionic Compounds Naming

Worksheet 6 - Naming Ionic Compounds Part 2 (Binary Compounds With Group 3 - 12 Metals) Naming (STOCK SYSTEM) ... AuP 8. Au 3 P 9. FeO 10. Sn 3 P 4 11. Fe 2 C 12. Co 4 C 3 13. Hg 3 N 14. MnS 2 15. AsI 5 16. SnO 17. Ti 2 S 3th, 2024

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- Is NOT An Ionic Compound; It Does Contain A Metal (Mn) And A Non-metal (O), But It Has A Charge. Thus, It Is A Polyatomic Ion, Not A Compound. A Compound Will NEVER Have A Charge! Naming Ionic Compounds There Are Three Steps Involved In Naming Ionic Compounds- Naming The Cation, Naming The Anion, And 3th, 2024

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