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CHAPTER 6:The Energy Balance For Chemical Reactors

$D(c A V R) Dt = R A V R$ Substituting In The Reaction-rate Expression, $R = K(T)c A$, And Using The Number Of Moles Of A, $N A = C A V R$ Yields $Dn A Dt = K(T)n A$ (6.20) Notice The Temperature Dependence Of $K(T)$ Prevents Us From Solving This Differential Equation Immediately. We Must Solve It Simultane 1th, 2024

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