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Form CO₂ And 0.50 G Of The Carbon Combusts To Form CO. The Resultant Mixture Of CO₂ And CO Is Passed Through Excess Na Feb 2th, 2024

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2) Aniline And Acetanilide Stand In A One-to-one Molar Relationship, So 0.0322 Mol Of Aniline Produces 0.0322 Mol Of Acetanilide. The Amount Of Acetic Anhydride Does Not Play A Role, Since It Is In Excess. Then, $n(\text{Aniline}) = n(\text{Acetanilide}) = 0.0322$ mol. We Find The Molar Mass Of Acetanilide: $M(\text{Acetanilide}) = 12 + 8 + 1 + 9 + 14 + 1 + 16 + 1 + 135 = 181$ g/mol. 89 M Aceta ... Jul 3th, 2024

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