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For A Monatomic Interacting Classical Gas, With Interactions That Only Depend On The Particle Co-ordinates, Derive The Maxwell Boltzmann Distribution Of Velocities And Show That The Average Kinetic Energy Is Given By $= \frac{3}{2} Nk_B T$. Solution. See Eqs. (94,95) Of The Notes. |||||{Quiz Problem 12. Using The Fact That $E = \frac{1}{2} m v^2 = \frac{1}{2} m \langle v^2 \rangle$ Show That $E = \frac{3}{2} Nk_B T$ Proportional $1 = N \frac{1}{2} m \langle v^2 \rangle = \frac{3}{2} Nk_B T$. Solution. See Eqs ... Jan 1th, 2024

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