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Four Of The Remaining 2p X And 2p Y Orbitals Must Make The  $\pi$  And  $\pi^*$  Bonding And Antibonding Orbitals. The "left Over" 2p X Orbital Goes Into An Approximately Nonbonding Orbital. (ie The Three 2p X Orbitals Form The Combinations +++, +.-, And +-+). (2 Marks For 2p X As B 1 And 2p Y As B 2, Both Of Which Should Be Read Straight From The Character Table) (d) How Many Valence Electrons ... 3th, 2024

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From The IR Rotational-vibrational Spectrum Of A Gas Phase Molecule, Can Measure: Bond Length Vibrational Force Constant Problem: For H81Br, Lines In The Above Spectrum Are Spaced By About 17 Cm-1. Calculate The Equilibrium Bond Length.  $h = 1.055 \times 10-34 \text{ J} \cdot \text{s} \cdot 1 \text{ Amu} = 1.661 \times 10-27 \text{ Kg} \cdot 3 \text{th}, 2024$ 

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