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27.0 The Standard Entropy Value For Steam Is A $+332 \text{ J K}^{-1} \text{ Mol}^{-1}$ B $+189 \text{ J K}^{-1} \text{ Mol}^{-1}$ C $+145 \text{ J K}^{-1} \text{ Mol}^{-1}$ D $+85 \text{ J K}^{-1} \text{ Mol}^{-1}$ (Total 1 Mark) Q25. Which One Of The Equations Below Represents A Reaction That Is Feasible ... 1th, 2024
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