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CALCULATIONS: 18. If 12.5 G Of Copper React With  
Excess Chlorine Gas, Then 25.4 G Of Copper(II)  
Chloride Are Produced. Find The Theoretical And  
Percent Yields.  $\text{Cu} + \text{Cl}_2 \rightarrow \text{CuCl}_2$  19. If 6.57 G Of Iron  
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Chapter 15 Solubility Worksheet SG 15.1 & 15.2  
Understanding Molarity Diluting Solutions Apr 2th,  
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Chapter 2 Atoms, Molecules, And Ions; Chapter 3  
Chemical Reactions And Reaction Stoichiometry:  
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Answers Calcium  $\text{C}_2\text{O}_2 + 3 \text{Cr}(3)$  When The Potassium

Chlorate ( $\text{KClO}_3$ ) Is Heated, It Decomposes To Form

Gases KCl And Oxygen ( $\text{O}_2$ .)  $2 \text{KClO}_3 \rightarrow 2 \text{KCl} + 3 \text{O}_2$  C.

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