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2.8 Formulas, Names And Masses Structural

Formula Of ...HCl As Ionic Or Covalent And Identify The Ions If Present. Problem: • Name The Compounds: PF<sub>5</sub>, CrF<sub>3</sub>, N<sub>2</sub>O<sub>3</sub>, Fe<sub>2</sub>O<sub>3</sub> - Molecular →phosphorus Pentafluoride - Ionic, F<sup>-</sup>, Cr<sup>3+</sup> →chromium(III) Fluoride - Molecular →dinitrogen Trioxide - Ionic, O<sub>2</sub><sup>-</sup>, Fe<sup>3+</sup> →iron(III) 1th, 2024Structural Formulas And Isomers2. The Compound C<sub>4</sub>H<sub>9</sub>OH Is An Isomer Of (1) C<sub>3</sub>H<sub>7</sub>COCH<sub>3</sub> (2) CH<sub>3</sub>COOC<sub>2</sub>H<sub>5</sub> (3) C<sub>2</sub>H<sub>5</sub>OC<sub>2</sub>H<sub>5</sub> (4) CH<sub>3</sub>COOH 3. If A Compound Has A Molecular Formula Of CH<sub>2</sub>O<sub>2</sub>, Then Its Structural Formula Must Be (1) (3) HO C O HHC O H O (2) (4) C O O H H HC O H O 4. The Structural Formulas HC CCCH H HHH H CCC H H C H H And H Represent Molecules Which Both Are (1) Halogen ...File Size: 128KBPage Count: 2 4th, 2024.

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Draw The Correct Structural FormulasFor Example Ch 3 Or Ch 2 Instead Of Showing All These Bonds, Draw A Correct Lewis Structure For Ch<sub>3</sub> 2chcooh See 35 The Two Ch<sub>3</sub> S Go At The Ends Vertically In The Middle Is The Linear HcoH With A Double Bonded O At The Second Carbon 36 Draw 1th, 2024

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