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$4 G = 6.67 \times 10^{-11} \text{ N}\cdot\text{m}^2\cdot\text{kg}^{-2}$ $\epsilon_0 = 8.85 \times 10^{-12} \text{ N}\cdot\text{m}^2\cdot\text{C}^{-2}$ $e = 1.6 \times 10^{-19} \text{ C}$
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Ion Source $V R^2 \epsilon_0 E_r M B V (\) \Rightarrow = | | ()$ This Is Regularly Used To Determine The Composition Of Unknown Samples. Particles With The Same Charge But Different Masses Strike The Plate In Different Locations. Example 21.16 -An α -particle Has A Charge Of $+2e$ And A Mass Of $6.64 \times 10^{-27} \text{ Kg}$. It Is Accelerated Jun 1th, 2024

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-2- CONSTANTS AND CONVERSION FACTORS Proton Mass, $1.67 \times 10^{-27} \text{ Kg}$ Neutron Mass, $1.67 \times 10^{-27} \text{ Kg}$ Electron Mass, $9.11 \times 10^{-31} \text{ Kg}$ Avogadro's Number, $6.02 \times 10^{23} \text{ mol}^{-1}$ Universal Gas Constant, $R = 8.31 \text{ J (mol K)}$

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