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Major Types Of Reactions Simplified By The Following  
Equations: Synthesis/Combination  $A + B \rightarrow AB$   
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Single Replacement  $X + AB \rightarrow XB + A$  Or  $Y + AB \rightarrow AY + B$   
Double Replacement  $XY + AB \rightarrow XB + AY$  ... Jan 3th,  
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 $2\text{FeCl}3$  B  $4\text{Fe}3\text{O}2$   $2\text{Fe}2\text{O}3$  C  $2\text{Fe}2\text{O}3$   $3\text{H}2\text{SO}4$   $1\text{Fe}2\text{SO}4$   
 $6\text{HBr}$  D  $1\text{C}4\text{H}6\text{O}3$   $1\text{H}2\text{O}$   $2\text{C}2\text{H}4\text{O}2$  E  $1\text{C}2\text{H}4$   $3\text{O}2$   $2\text{CO}2$   $2\text{H}2\text{O}$   
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